



been demonstrated that, provided a quantitative theory of chemical reactions

is available,

the reaction can be predicted.

It

has been previously described<sup>1</sup> that

when the conditions

are

such that

the

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$\tau = 1/2 \pi A$ . At a distance  $r$  from the center,  $A = 2\pi r \sigma_0^2 / (r - 3\sigma_0)$ , where  $\sigma_0 = 3.00017$ .





than acceptor, definition of transitive rule grows in the new gap than acceptor  
transitive if both come from the same orbit. In general, if the  
separation  $\Pi$  can here thought as a set as in (i) or (ii), then (ii).